How does salt water intrusion affect coupled iron and sulfur cycling in a coastal freshwater wetland?

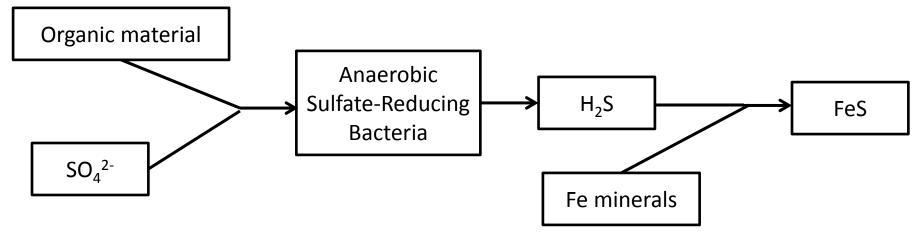
Valerie Schoepfer, Amy Burgin, Emily Bernhardt University of Nebraska, Duke University



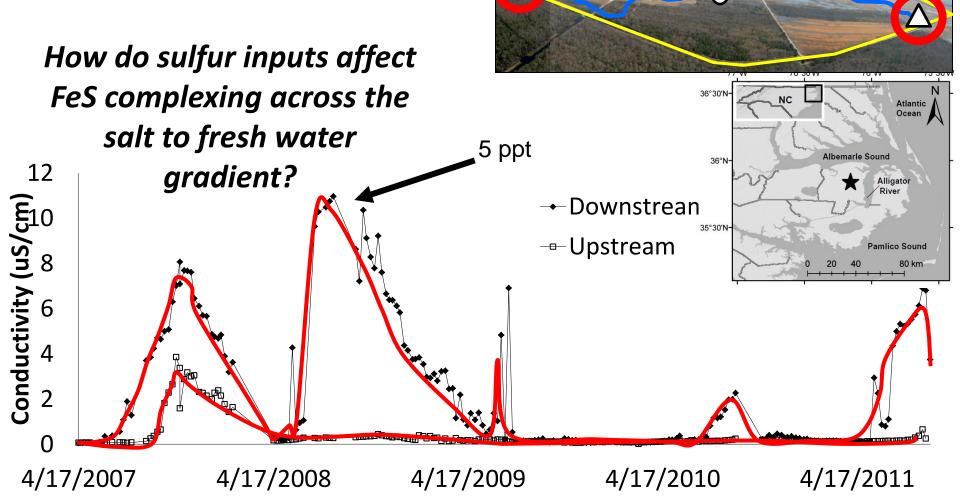


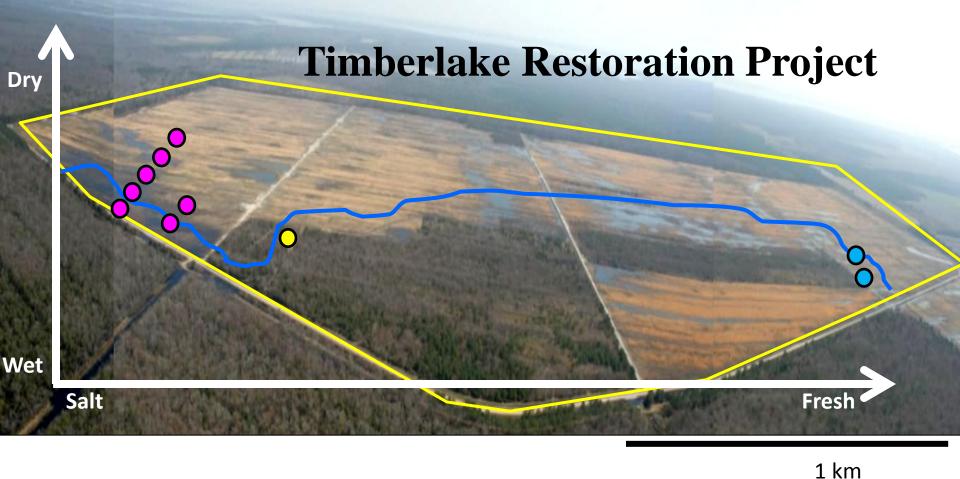
Climate change, sulfur and iron: implications for coastal freshwater wetlands

- Salt water is a large source of sulfate
- Microbes reduce sulfate to sulfide
 - Toxic to many organisms
 - With salt, organizes coastal communities
- Reduced iron binds with sulfide



Periodic Salt Water Intrusion





Seasonal intrusion depending - 440 ha on groundwater levels - 10 sites

Salt water intrusion as a natural sulfur addition

Indicators of Sulfate Reducing Activity

Traditional ³⁵S

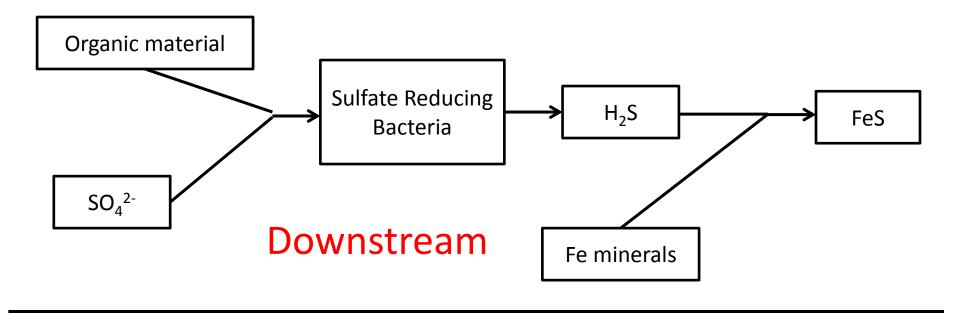
- Radioisotope
- Uses soil core (5cm diameter)
- Glovebag
- Few samples (expensive and time consuming)
- Proven method

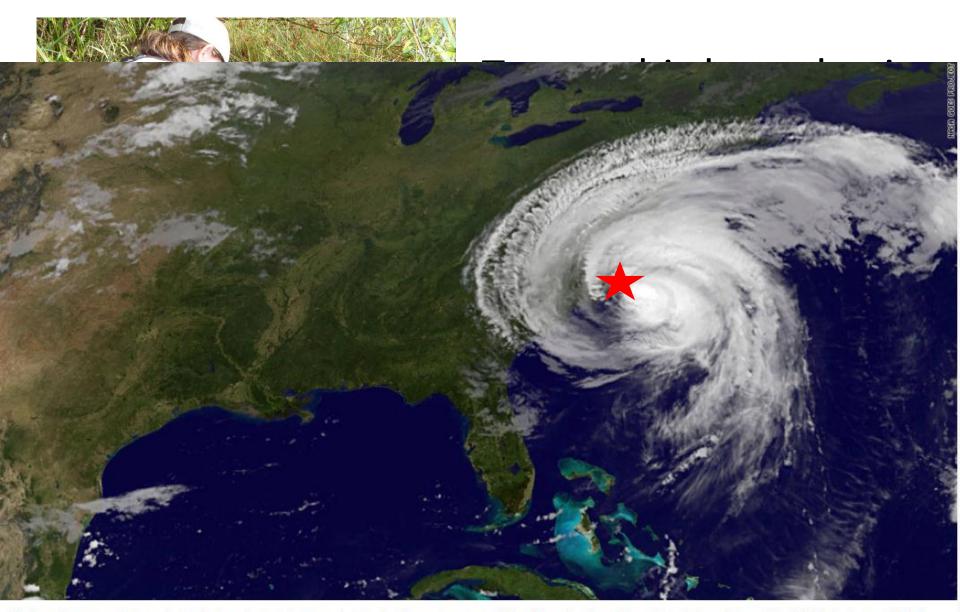
Indicator of Reduction in Soils Method (IRIS, Rabenhorst et al 2010)

- Increased SA (20x15cm)
- Efficient, inexpensive
- Deploy easily
- See biogeochemistry in action
- New method



Predictions

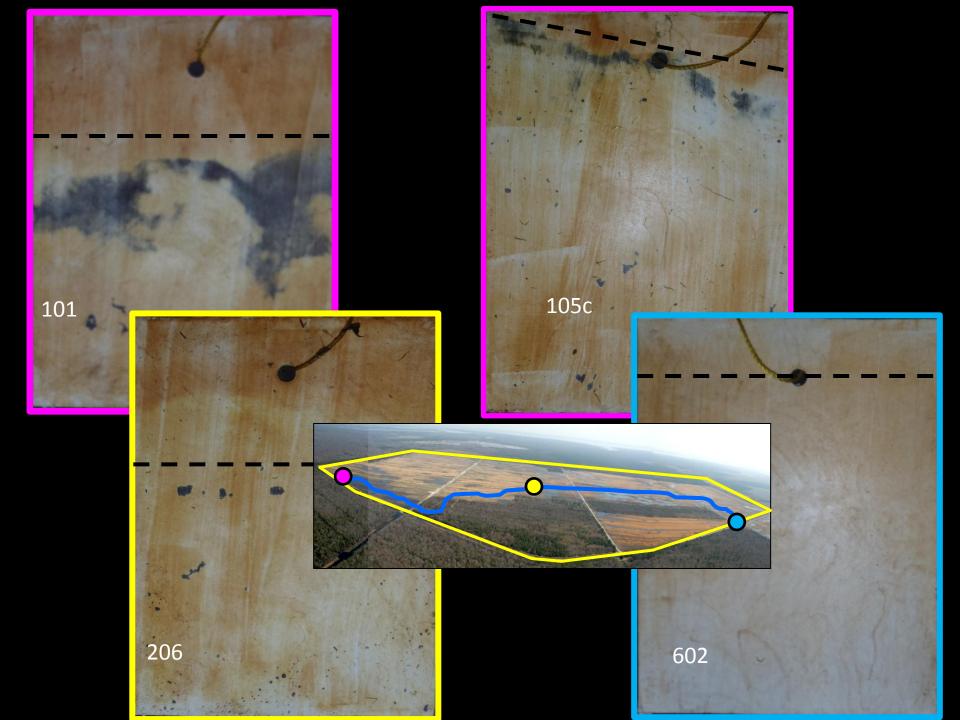




Hurricane Irene was downgraded early Saturday to a Category 1 storm, but forecasters warned it's still a major threat. Ferocious winds and torrential rainfall lashed coastal North





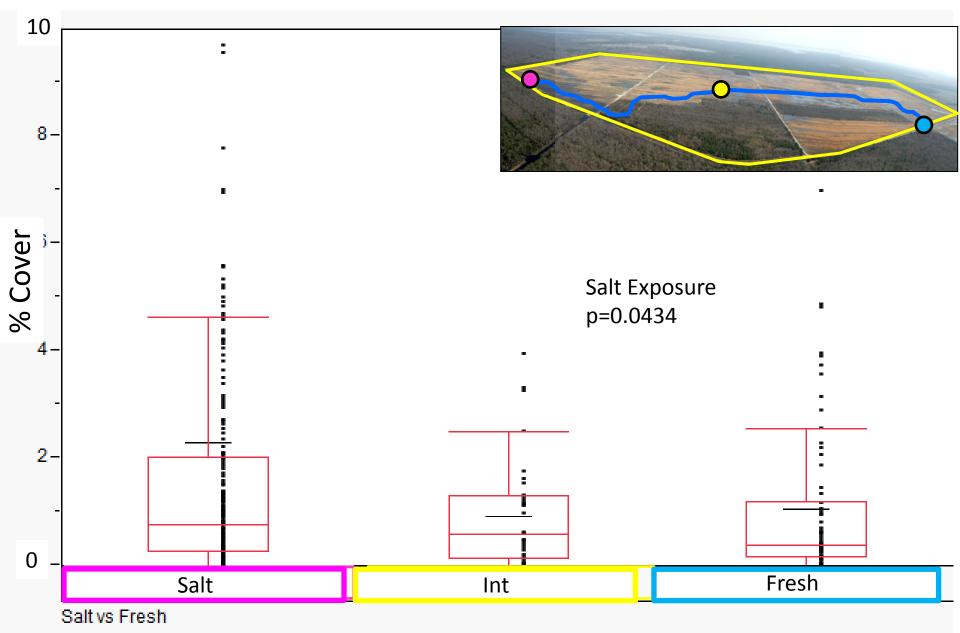


June sulfate reduction rate (³⁵S) along the saline to fresh gradient

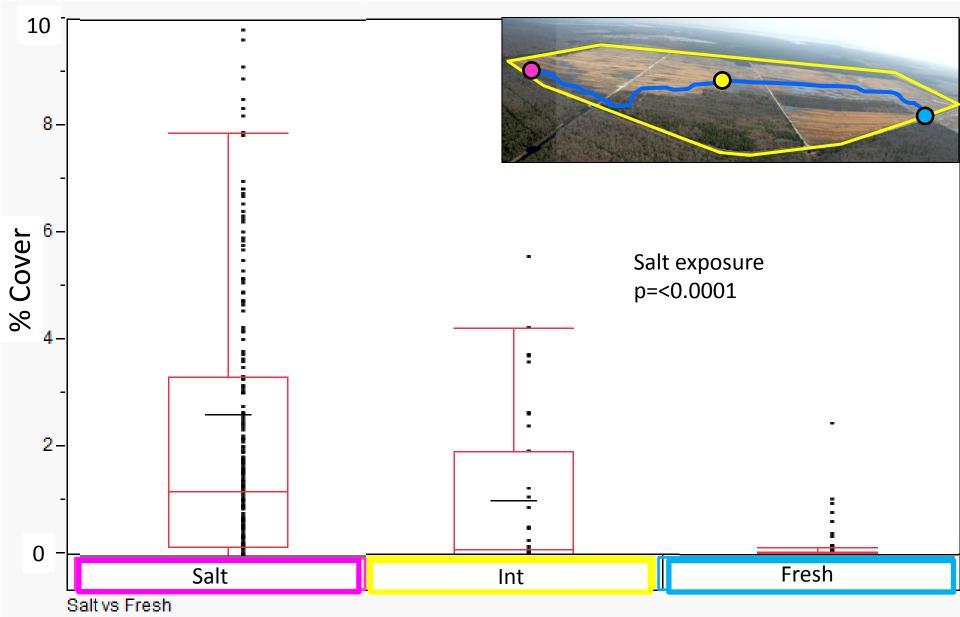


SRR (umoles/cm3/hr) Salt exposure p=0.778 0-Fresh Salt Int Salt to Fresh

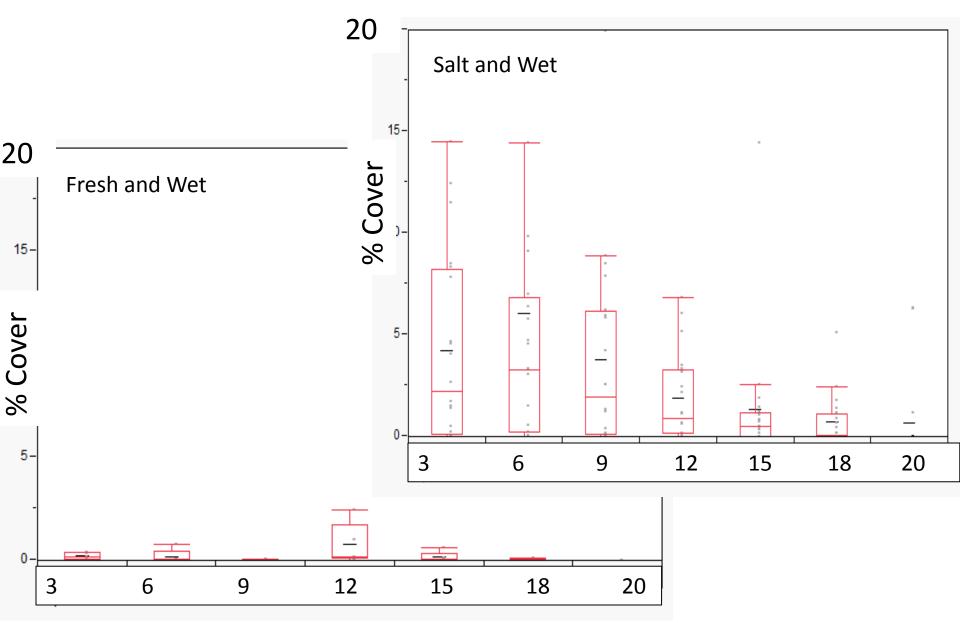
IRIS June Percent Cover



IRIS September Percent Cover



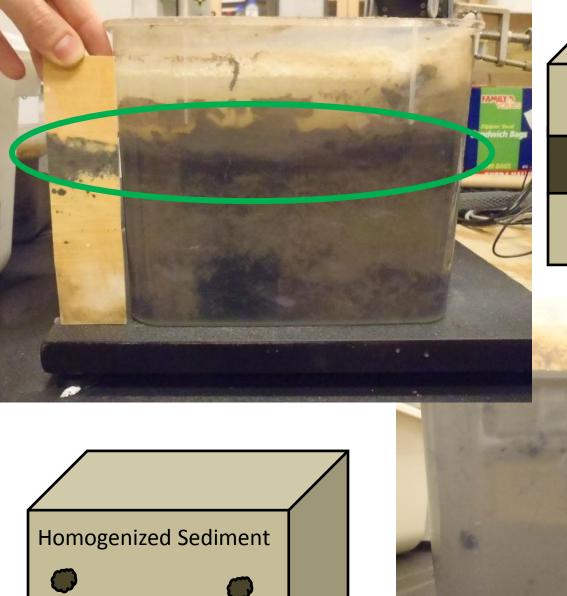
September depth gradient



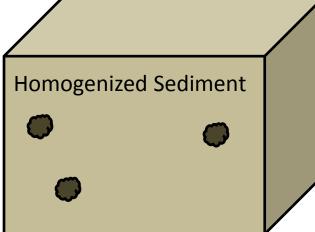
Analyzing the photos

- Percent cover
 (abundance of dark areas)
- Intensity of darkness (concentration)
- Why pockets vs bands?





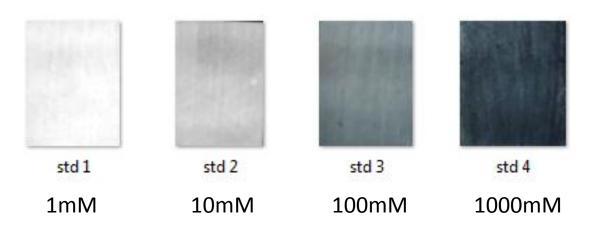
Dextrose and Sulfate





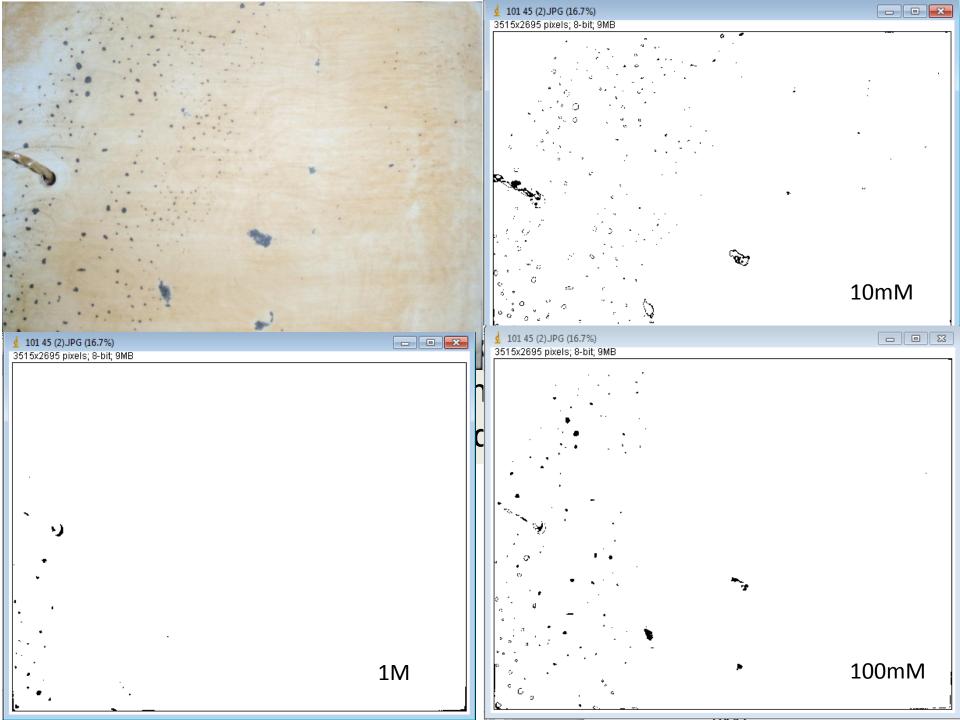
Future Work

• Improve photography to obtain darkness of compounds and therefore quantitative rates

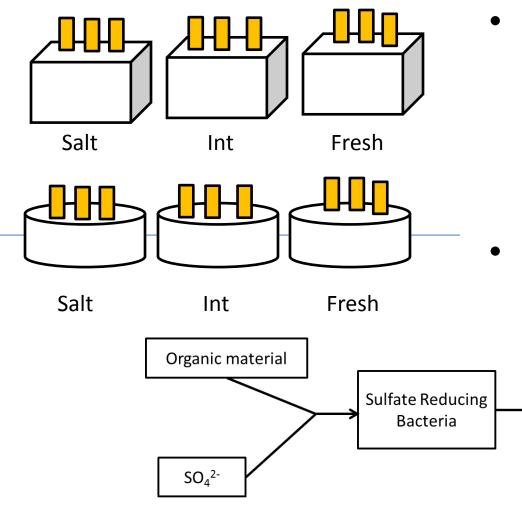


- Sulfide solution in plastic bag
- Held in glovebox for 24 hours





Meso- and Microcosm Experimental Design



- Water amendments
 - (0, 2.5, 5ppt)

 H_2S

Fe minerals

Lab and field component

FeS

- 3 treatments, 2 sites, 3 reps (18 total)
- Compare plates to ³⁵S

Conclusions

- Sulfate reduction increases with intrusion, and is greater in wetter sites
- Sea level rise and periodic salt water intrusion will increase coastal sulfate reduction
 - Impacts of sulfate reduction are heavily mediated by iron availability
- How much exposure is necessary to switch to sulfidic conditions?

Acknowledgements



Indicator of Reduction in Soils Method

(IRIS, Rabenhorst et al. 2010)

- Iron oxide paint on PVC
- Inserted into sediment for 24 hours
- Fe on plate binds with porewater sulfide
 - Black complexes that fade with exposure to O₂
- See biogeochemistry in action
 - Visual reaction with sulfide and a way to quantify sulfide concentrations

